

# PERCENTAGE ERROR

Name \_\_\_\_\_

Percentage error is a way for scientists to express how far off a laboratory value is from the commonly accepted value.

The formula is:

$$\% \text{ error} = \frac{\left| \text{Accepted Value} - \text{Experimental Value} \right|}{\text{Accepted Value}} \times 100$$

→  
absolute value

Determine the percentage error in the following problems.

1. Experimental Value = 1.24 g  
Accepted Value = 1.30 g

Answer: \_\_\_\_\_

2. Experimental Value =  $1.24 \times 10^{-2}$  g  
Accepted Value =  $9.98 \times 10^{-3}$  g

Answer: \_\_\_\_\_

3. Experimental Value = 252 mL  
Accepted Value = 225 mL

Answer: \_\_\_\_\_

4. Experimental Value = 22.2 L  
Accepted Value = 22.4 L

Answer: \_\_\_\_\_

5. Experimental Value = 125.2 mg  
Accepted Value = 124.8 mg

Answer: \_\_\_\_\_